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Student Exploration

Temperature And Particle

The Temperature and Particle

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Motion Gizmo™ illustrates how the molecules of gas move at different temperatures. In this Gizmo, temperature is measured on the Kelvin scale, which measures temperature from absolute zero, the coldest possible temperature ($-273.15\text{ }^{\circ}\text{C}$).

*Student Exploration:
Temperature and Particle
Motion*

Student Exploration:
Temperature and Particle
Motion Question: How is the
temperature of a gas related
to the motion of gas
molecules? 1. Observe: Move
the Temperature slider back
and forth. Focus on the

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Motion Answers

particle motion at left.
What do you notice? The
colder it gets the slower
they go the hotter it gets
the faster they will go.

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Particle Motion.docx -
Student ...*

Gizmo Warm-up The
Temperature and Particle
Motion Gizmo™ illustrates
how the molecules of gas
move at different
temperatures. In this Gizmo,
temperature is measured on
the Kelvin scale, which
measures temperature from
absolute zero, the coldest
possible temperature
(-273.15 °C).

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*Student Exploration-Particle
Temperature and Particle
Motion ...*

Name: Anaya Tei Date:
October 23,2020 Student
Exploration: Temperature and
Particle Motion Vocabulary:
absolute zero, Kelvin scale,
kinetic energy, Maxwell-
Boltzmann distribution,
molar mass, molecule,
temperature, universal gas
constant Prior Knowledge
Questions (Do these BEFORE
using the Gizmo.) 1. Why is
hot air hot? Hot air is hot
because the sun is radiating
hot oxygen 2.

*Science .pdf - Name Anaya
Tei Date October 23,2020
Student ...*

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The Temperature And Particle Motion Gizmo™ illustrates how the molecules of gas move at different temperatures. In this Gizmo, temperature is measured on the Kelvin scale, which measures temperature from absolute zero, the coldest possible temperature (-273.15 °C).

*Student Exploration:
Temperature And Particle
Motion | pdf ...*

2019 Name: _____ Date: _____

Student Exploration:
Temperature and Particle
Motion Vocabulary: absolute
zero, Kelvin scale, kinetic
energy, Maxwell-Boltzmann
distribution, molar mass,

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Temperature and Particle
Motion Answers
molecule, temperature, universal gas constant Prior
Knowledge Questions (Do these BEFORE using the
Gizmo.) 1.

Temperature_and_Particle_Motion_Gizmo.docx - Name Date

...

Student Exploration:
Temperature and Particle
Motion 4 Prior Knowledge
Questions (Do these BEFORE
using the Gizmo.) 1. Why is
hot air hot? Hot air rises
because when you heat air
(or any other gas for that
matter), it expands. When
the air expands, it becomes
less dense than the air
around it.

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Copy of R Temperature and Particle Motion.docx - Student ...

Student Exploration:
Temperature and Particle Motion. Vocabulary: absolute zero, Kelvin scale, kinetic energy, Maxwell-Boltzmann distribution, molar mass, molecule, temperature, universal gas constant.
Prior Knowledge Questions (Do these BEFORE using the Gizmo.) Why is hot air hot?

Temperature and Particle Motion

In the Temperature and Particle Motion Gizmo, students explore how the temperature and molecular

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weight of a gas relates to the distribution of particle velocities. The Gizmo includes a simulation that shows how particles in a gas collide and how momentum and kinetic energy are transferred between particles.

*Gizmo of the Week:
Temperature and Particle
Motion ...*

Temperature and Particle Motion Observe the movement of particles of an ideal gas at a variety of temperatures. A histogram showing the Maxwell-Boltzmann velocity distribution is shown, and the most probable velocity,

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mean velocity, and root mean square velocity can be calculated. Molecules of different gases can be compared.

*Temperature and Particle
Motion Gizmo : Lesson Info*

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Temperature and Particle
Motion Gizmo™ illustrates
how the molecules of gas
move at different
temperatures. In this Gizmo,
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the Kelvin scale, which
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Motion The Temperature and
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illustrates how the
molecules of gas move at
different temperatures. In
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measured on the Kelvin
scale, which measures
temperature from absolute
zero, the coldest possible
temperature ($-273.15\text{ }^{\circ}\text{C}$).

*Temperature And Particle
Motion Gizmo Answer Key |
www . . .*

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Temperature and Particle Motion The Temperature and Particle Motion Gizmo²¹²² illustrates ... of the curve and your answer to the previous question, do you expect the mean velocity to [Filename:

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scale, which measures temperature from absolute zero, the coldest possible temperature ($-273.15\text{ }^{\circ}\text{C}$).

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Temperature and Particle
Motion Gizmo™ illustrates
how the molecules of gas
move at different
temperatures. In this Gizmo,
temperature is measured on
the Kelvin scale, which
measures temperature from
absolute zero, the coldest
possible temperature
(-273.15 °C).

Solubility And Temperature

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Gizmo Answer Key Activity A

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